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# Chloride - Single Reagent

Diagnostic reagent for the in-vitro quantitative determination of Chloride in human serum, plasma and urine on manual systems.

REF: BS.1/CL02.025.0050	50 test	REF: BS.1/CL02.050.0100	100 test
REF: BS.1/CL04.025.0100	100 test	REF: BS.1/CL02.100.0200	200 test

# **CLINICAL SIGNIFICANCE**

Chloride is the most abundant extracellular anion. Together with sodium, chloride is responsible for the maintenance of osmotic pressure, the anion-cation balance and therefore of the water distribution in the extracellular fluid compartment.

Decreased plasma Cl<sup>-</sup> concentrations (hypochloremia) can result

rom salt-losing nephritis, persistent gastric secretion, prolonged vomiting and metabolic acidosis that are caused by increased production or reduced secretion of organic acids. Increased plasma CI<sup>-</sup> concentrations (hyperchloremia) occur with dehydration, renal tubular acidosis, acute renal failure, in adrenocortical hyperfunction, salicylate in-toxication and metabolic acidosis associated with prolonged diarrhoea and loss of sedium bicarbonate. of sodium bicarbonate.

## **METHOD PRINCIPLE**

Chloride ions react with undissociated mercuric thiocyanate to form undissociated mercuric Chloride and free thiocyanate ions. thiocyanate ions react with ferric ions to form a highly coloured reddish complex of ferric thiocyanate.

The increasing extinction is directly proportional to the concentration of chloride ions.

→ HgCl2 + 2 SCN<sup>-</sup> Hg (SCN)2 + 2CI-3(SCN) - + Fe 3+ Fe(SCN)3

# REAGENT COMPOSITION

R1: Standard	354.6 mg/dl (100 mmol/L)
R2: Reagent Hg II - thiocyanate Fe III - nitrate HNO3	2 mmol/L 30 mmol/L 40 mmol/L

# PRECAUTIONS AND WARNINGS

Reagent to be handled by entitled and professionally educated person.

Good Laboratories practices using appropriate precautions should be followed in:

- · Wearing personnel protective equipment (overall, gloves, glasses,).
- Do not pipette by mouth.
- · In case of contact with eyes or skin; rinse immediately with plenty of soap and water. In case of severe injuries; seek medical advice immediately.
- Respect country requirement for waste disposal.
  - \$56: dispose of this material and its container at hazardous or special waste collection point.

S57: use appropriate container to avoid environmental contamination.

S61: avoid release in environment.

For further information, refer to the Chloride reagent material safety data sheet.

#### REAGENT PREPARATION, STORAGE AND STABILITY

BioScien Chloride reagents are supplied ready-to-use. All reagents are stable until expiration date stated on label when stored at 2-8°C.

#### Deterioration

The BioScien Chloride reagent is normally clear, do not use reagent if it is turbid.

## SPECIMEN COLLECTION AND PRESERVATION

Unhemolysed serum or Heparinized plasma or urine. Chloride is stable for 7 days at 2-8°C.

#### SYSTEM PARAMETERS

Wavelength	492 nm (460 - 500 nm)
Optical path	1 cm
Assay type	End-point
Direction	Increase
Sample Reagent Ratio	1:100
e.g.: Reagent volume	1 ml
Sample volume	10 µl
Temperature	15-25°C
Incubation time	5 min at 15-25°C
Zero adjustment	Reagent blank
Sensitivity	1.0 mmol/L
Linearity	130 mmol/L

## **EQUIPMENT REQUIRED NOT PROVIDED**

- · Sterile Syringe
- Analytical tubes, automatic pipet
- · Centrifuge and spectrophotometer

## **ASSAY PROCEDURE**

	Blank	Standard	Specimen
Reagent	1.0 ml	1.0 ml	1.0 ml
Standard		10 µl	
Specimen			10 µl

Mix, incubate for 5 minutes at room temp. Measure absorbance of specimen and standard against reagent blank. The color is stable for at least 30 minutes.

## CALCULATION

Chloride concentration (mmol/L) = Abs. specimen x 100 Abs. standard

# **QUALITY CONTROL**

To ensure adequate quality control, it is recommended that normal and abnormal commercial control serum of known concentrations included in each run. If control values are found outside the defined range, check the instrument calibration, and reagent for problems. If control still out of range please contact BioScien technical support.

## PERFORMANCE CHARACTERISTICS

Precision	Within run		Run to run	
	(Repeatability)		(Reproducibility)	
	Normal level	High level	Normal level	High level
n	20	20	20	20
Mean mmol/L	85.1	136.3	84.6	133.4
SD. mmol/L	0.51	0.94	0.67	1.95
CV. %	0.60	0.69	0.79	1.5

The results of the performance characteristics depend on the analyzer used.

# Accuracy (Methods Comparison)

Result obtained from **BioScien** Chloride reagent compared with commercial reagent of the same methodology performed on 20 human sera give a correlation of 0.994.

#### Sensitivity

When run as recommended, the minimum detection limit of the assay is 1.0  $\mbox{mmol}\slash\!\!/ \mbox{L}.$ 

#### Linearity

The reaction is linear up to concentration of 130 mmol/L. Specimens showing higher concentration should be diluted 1+1 using physiological saline and repeat the assay (resultx2).

#### INTERFERING SUBSTANCES

## Haemoglobin

It interferes with Over-estimation value.

#### Icterus

No significant interference.

## lipemia

No significant interference.

# Bromide and Fluoride

They can cause falsely elevated chloride values.

#### Drugs

No interference was observed by ascorbic acid up to 10 mg/dl.

#### Others

Other drugs and substances may interfere.

# **EXPECTED VALUES**

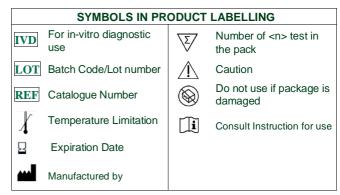
	mmol/l
Serum / Plasma	97 – 108
Urine(24 h)	95 - 240 mmol/24h
Urine(morning)	54 – 158

# **DYNAMIC RANGE**

1.0-130 mmol/l.

# **REFERENCES**

- 1.Tietz N.W. Clinical Guide to Laboratory Tests, 3rd Philadelphia:W.B. Saunders Company, 1995:516-519.
- Batlle DC. et al. The use of the urinary anion gap in the diagnosis of hyperchloremic metabolic acidosis. N Engl J Med 1988, 318:594-599.
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- 4.YOUNG D.S., Effect of Drugs on Clinical Laboratory Tests,4<sup>th</sup> Ed. (1995) p. 3-137 a 3-141.





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